

Q1. Which of the following is NOT a mirror formula?

- a) $1/f = 1/v + 1/u$
- b) $m = v/u$
- c) $m = -v/u$
- d) $f = v + u$

Ans:- d

Q2. The splitting of white light into seven colors on passing through a prism is called:

- a) Scattering b) Dispersion c) Refraction d) Reflection

Q3. A person cannot see nearby objects clearly. The defect is:

- a) Myopia b) Hypermetropia c) Presbyopia d) Astigmatism

Ans:

Q4. Which of the following is NOT a physical change?

- a) Melting of ice
- b) Rusting of iron
- c) Boiling of water
- d) Dissolution of salt

Ans:

Q5. Photosynthesis occurs in:

- a) Mitochondria b) Chloroplast c) Ribosome d) Nucleus

Q6. Which organelle is called the powerhouse of the cell?

- a) Nucleus b) Mitochondria c) Chloroplast d) Ribosome

Q7. The process of removal of undigested food is:

- a) Excretion b) Egestion c) Respiration d) Assimilation

Q8. Which blood vessel carries blood from lungs to heart?

- a) Pulmonary artery b) Pulmonary vein c) Aorta d) Vena cava

Q9. Which of the following is NOT a plant hormone?

- a) Auxin b) Insulin c) Gibberellin d) Cytokinin

Q10. Which part of the brain controls voluntary actions?

- a) Medulla b) Cerebellum c) Cerebrum d) Pons

Q11. Which plant hormone promotes cell elongation?

- a) Cytokinin b) Auxin c) Abscissic acid d) Ethylene

Q12. Reflex actions are controlled by:

- a) Brain b) Spinal cord c) Cerebellum d) Medulla

Q13. The electrical impulse in nerves is carried by:

- a) Hormones b) Neurotransmitters c) Dendrites d) Myelin sheath

Q13. Which hormone regulates sugar level in blood?

- a) Thyroxine b) Adrenaline c) Insulin d) Testosterone

Q14. The basic functional unit of the kidney is:

- a) Nephron b) Alveolus c) Neuron d) Glomerulus

Q15. Which enzyme is present in saliva?

- a) Lipase b) Ptyalin c) Pepsin d) Trypsin

Q16. In photosynthesis, light energy is absorbed by:

- a) Chlorophyll b) Mitochondria c) Cytoplasm d) Vacuole

Q17. The site of exchange of gases in human lungs is:

- a) Trachea b) Bronchi c) Alveoli d) Diaphragm

Q18. A base which is soluble in water is called:

- a) Alkali b) Salt c) Neutral base d) Strong base

Q19. Which acid is present in ant sting?

- a) Acetic acid b) Methanoic acid c) Citric acid d) Lactic acid

Q20. When CO_2 is passed through lime water, it turns milky due to:

- a) CaSO_4 b) CaCO_3 c) CaCl_2 d) Ca(OH)_2

Q21. Which gas is evolved when a metal carbonate reacts with an acid?

- a) H_2 b) CO_2 c) O_2 d) Cl_2

Q22. Tooth decay starts when pH of mouth falls below:

- a) 7.4 b) 6.5 c) 5.5 d) 4.5

Q23. The pH of NaOH solution is around:

- a) 1 b) 7 c) 10-12 d) 14

Q24. Which of the following salts is formed when hydrochloric acid reacts with sodium hydroxide?

- a) Na_2CO_3 b) NaCl c) NaNO_3 d) Na_2SO_4

Q25. Rust is chemically:

- a) FeO b) Fe_2O_3 c) $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ d) Fe(OH)_2

Q26. Which of the following prevents rusting of iron?

- a) Painting b) Galvanisation c) Alloying d) All of these

Q27. Brass is an alloy of:

- a) $\text{Cu} + \text{Sn}$ b) $\text{Cu} + \text{Zn}$ c) $\text{Fe} + \text{C}$ d) $\text{Fe} + \text{Cr} + \text{Ni}$

Q28. Stainless steel contains:

- a) $\text{Fe} + \text{Zn}$ b) $\text{Fe} + \text{C} + \text{Ni} + \text{Cr}$ c) $\text{Fe} + \text{Sn}$ d) $\text{Fe} + \text{Cu}$

Q29. The phenomenon of depositing a layer of zinc on iron to prevent rusting is called:

- a) Alloying b) Electroplating c) Galvanisation d) Annealing

Q30. Which method is used for refining of highly reactive metals like sodium?

- a) Distillation b) Electrolysis c) Liquefaction d) Zone refining

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Q31. Which gas is evolved when zinc reacts with dilute HCl?

- a) Oxygen b) Hydrogen c) Nitrogen d) Carbon dioxide

Q32. The amphoteric oxide among the following is:

- a) Na_2O b) Al_2O_3 c) CO_2 d) MgO

Q33. Which metal does not react with cold or hot water but reacts with steam?

- a) Na b) K c) Zn d) Ca

Q34. Non-metal oxides are generally:

- a) Acidic b) Basic c) Amphoteric d) Neutral

Q35. The blood vessel that carries oxygenated blood from lungs to heart is:

- a) Pulmonary artery b) Pulmonary vein c) Vena cava d) Aorta

Q36. The universal donor blood group is:

- a) A b) B c) AB d) O

Q37. Xylem transports:

- a) Water and minerals b) Food c) Hormones d) Enzymes

Q38. Phloem transports food in plants by:

- a) Transpiration pull b) Active transport c) Osmosis d) Diffusion

Q39. Functional unit of kidney is:

- a) Nephron b) Alveolus c) Glomerulus d) Tubule

Q40. In plants, the main excretory product is:

- a) Oxygen b) Carbon dioxide c) Water d) All of these

Q41. Which part of nephron filters blood?

- a) Collecting duct b) Tubule c) Bowman's capsule d) Loop of Henle

Q42. Which of the following is not a life process?

- a) Nutrition b) Respiration c) Excretion d) Thinking

Q43. Which defect of vision is corrected by a concave lens?

- a) Myopia
b) Hypermetropia
c) Presbyopia
d) Cataract

Q44. Which defect of vision is corrected using a convex lens?

- a) Myopia
b) Hypermetropia
c) Both myopia and hypermetropia
d) None

Q45. Which phenomenon is responsible for twinkling of stars?

- a) Reflection
b) Refraction
c) Dispersion
d) Diffraction

Q46. The splitting of white light into seven colours by a prism is called:

- a) Refraction
b) Dispersion
c) Scattering
d) Interference

Q47. The sky appears blue due to:

- a) Dispersion
b) Scattering of light
c) Reflection
d) Refraction

Q48. The danger signal lights are red because:

- a) Red is most attractive
b) Red light is least scattered
c) Red light has shortest wavelength
d) Red is a primary colour

Q49. Myopia can be corrected by:

- a) Convex lens
b) Concave lens
c) Cylindrical lens
d) Bifocal lens

50. Which of the following is a decomposition reaction?

- a) $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$
b) $2\text{HgO} \rightarrow 2\text{Hg} + \text{O}_2$
c) $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
d) $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

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The brown coating formed on iron on long exposure to moisture is:

- a) $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$
- b) FeO
- c) Fe_2O_3
- d) $\text{Fe}(\text{OH})_2$

Which of the following is NOT a physical change?

- a) Evaporation of water
- b) Melting of ice
- c) Rusting of iron
- d) Dissolution of sugar

The reaction: $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ is:

- a) Combination
- b) Decomposition
- c) Displacement
- d) Neutralisation

Which of the following shows both oxidation and reduction?

- a) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
- b) $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$
- c) $\text{Fe} + \text{S} \rightarrow \text{FeS}$
- d) $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$

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